### POZNAN UNIVERSITY OF TECHNOLOGY



#### EUROPEAN CREDIT TRANSFER AND ACCUMULATION SYSTEM (ECTS)

general academic

Course offered in

compulsory

pl. M. Skłodowskiej-Curie 5, 60-965 Poznań

# **COURSE DESCRIPTION CARD - SYLLABUS**

Course name

Język angielski specjalistyczny (Specialist English)

**Course** 

Field of study Year/Semester

Technologia chemiczna (Chemical Technology) I/1

Area of study (specialization) Profile of study

Technologia chemiczna ogólna (General chemical technology)

Level of study

Second-cycle studies Polish

Form of study Requirements

**Number of hours** 

part-time

Lecture Laboratory classes Other (e.g. online)

0 0

Tutorials Projects/seminars

20 0

**Number of credit points** 

2

#### **Lecturers**

Responsible for the course/lecturer: Responsible for the course/lecturer:

Dorota Żarnowska M.Sc. eng

## **Prerequisites**

The already acquired language competence compatible with level B2 (CEFR)

The ability to use general and field specific vocabulary, and grammatical structures required on the first level of studies

The ability to work individually and in a group; the ability to use various sources of information and reference works.

## **Course objective**

- 1. Advancing students' language competence towards the level at least B2+ (CEFR).
- 2. Development of the ability to use academic and field specific language effectively in both receptive and productive language skills.
- 3. Improving the ability to understand field specific texts (familiarizing students with basic translation techniques).
- 4. Improving the ability to function effectively on an international market and on a daily basis.

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### **Course-related learning outcomes**

#### Knowledge

As a result of the course, the student ought to acquire field specific vocabulary related to the following issues:

- 1. English for academic purposes an abstract
- 2. Electrolysis
- 3. Electroplating
- 4. Petrochemicals and polymers
- 5. Addition and condensation polymerisation

and also define and explain notions, phenomena and processes connected with them.

K\_W03, K\_W06, P7S\_WG

#### Skills

As a result of the course, the student is able to:

- give a talk on field specific or popular science topic (in English)
- discuss general and field specific issues using an appropriate linguistic and grammatical repertoire,
- formulate a text in English where he/she explains/describes a selected field specific topic, understand and analyze international, field specific literature
- give a presentation of his/her engineering thesis

K U01, K U03, K U06, P 7SUK

## Social competences

As a result of the course, the student is able to communicate effectively in a field specific/professional area, and to give a successful presentation in English.

He/she is able to recognize and understand cultural differences in a professional and private conversation, and in a different cultural environment.

K\_K01, K\_K03, K\_K06, P\_7SKK

# Methods for verifying learning outcomes and assessment criteria

Learning outcomes presented above are verified as follows:

- Formative assessment: tests during academic year (written and oral) presentations
- Summative assessment: credit

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### **Programme content**

- 1. English for academic purposes an abstract
- 2. Electrolysis
- 3. Electroplating
- 4. Petrochemicals and polymers

### **Teaching methods**

work with texts, discussion, team work, translation, films, individual written and oral deliverance, individual meetings with students, homework analysis, classes on e-meeting platform, Moodle platform exercises...

## **Bibliography**

#### **Basic**

Richard Harwood and Ian Lodge, Cambridge IGCSE Chemistry, Coursebook, Fourth edition, 2014, Cambridge University Press, (IGCS)

Urszula Kamińska, English for Biotechnology, 2016, Publishing House, Gdańsk University of Technology

English for Academics Book 1, Cambridge, British Council

### Additional

Richard Harwood and Ian Lodge, Cambridge IGCSE Chemistry, Workbook, Fourth edition, 2014, Cambridge University Press, (IGCS -W)

Gallagher, Rose Marie and Ingram, Paul. 2011. Complete Chemistry. Oxford: Oxford University Press

# Breakdown of average student's workload

	Hours	ECTS
Total workload	50	2,0
Classes requiring direct contact with the teacher	25	1,0
Student's own work (literature studies, preparation for tutorials,	25	1,0
preparation for tests) 1		

<sup>&</sup>lt;sup>1</sup> delete or add other activities as appropriate